

Seven ABG steps, by Alex Flaxman, MD, MSE

1. Calc A-a gradient =  $150 - [1.25 \times PaCO_2] - PaO_2$
2. Alk or acid
3. 1° disturb resp or metab
4. if resp, ac or ch

Expected pH chg

$$\text{ac resp acid pH decrease} = .08(pCO_2 - 40) / 10$$

$$\text{ch resp acid pH decrease} = .03(pCO_2 - 40) / 10$$

$$\text{ac resp alk pH increase} = .08(40 - pCO_2) / 10$$

$$\text{ch resp alk pH increase} = .03(40 - pCO_2) / 10$$

Expected bicarb

$$\uparrow \text{ ac resp acid } .1 (pCO_2 - \text{norm } pCO_2)$$

$$\uparrow \text{ ch resp acid } .35 (pCO_2 - \text{norm } pCO_2) \quad \text{some say coefficient } 0.4$$

$$\downarrow \text{ ac resp alk } .2 (\text{norm } pCO_2 - pCO_2)$$

$$\downarrow \text{ ch resp alk } .5 (\text{norm } pCO_2 - pCO_2)$$

Can also use

Resp acidosis Ac for ea 10 ↑ of pCO<sub>2</sub>, HCO<sub>3</sub> ↑ by 1

Ch for ea 10 ↑ of pCO<sub>2</sub>, HCO<sub>3</sub> ↑ by 2

Resp alk Ac for ea 10 ↓ of pCO<sub>2</sub>, HCO<sub>3</sub> ↓ by 1

Ch for ea 10 ↓ of pCO<sub>2</sub>, HCO<sub>3</sub> ↓ by 5

$$\text{Metab acid } pCO_2 = (1.5 \times HCO_3) + 8 \pm 2$$

Another way: for ea 1 ↓ in bicarb from 24, pCO<sub>2</sub> ↓ by 1

$$\text{Metab alk} = \text{for ea } \uparrow \text{ in } HCO_3 \text{ by } 1, pCO_2 \uparrow \text{ by } \frac{1}{2}$$

5. If metab, is there an AG?  $Na - (Cl + HCO_3)$ 
  - For every 1 g/dL decrease in albumin below 4, the AG should be raised by 2.5 mEq/L
  - (this is generally not on med school- or residency-level exams)
6. If AG metab acidosis exists, is there another metab proc going on? Use  $\delta/\delta$  or corrected HCO<sub>3</sub>.
7.  $\delta/\delta = \text{chg in AG} / \text{chg in } HCO_3 = (AG - 12) / (24 - HCO_3)$ 

For med school- and residency-level exams:

  - $\delta/\delta > 1 \rightarrow$  concurrent metab alk
  - $\delta/\delta < 1 \rightarrow$  concurrent non-AG (hyperchloremic) metab acid

< 0.4 then there is a concomitant hyperchloremic normal anion gap acidosis

0.4-0.8 then consider combined high AG and normal AG acidosis but ratio is often < 1 in acidosis associated with renal failure

1-2 is usual for uncomplicated high AG acidosis

Lactic acidosis yields an average value of 1.6

DKA yields value closer to 1 b/c of urine ketone loss (espec if pt not dehydrated)

If > 1 or definitely 2 then concurrent metabolic alkalosis (maybe a pre-existing one?)

$$[\text{corrected } HCO_3 = \text{measured } HCO_3 + AG - 12]$$

-6  $\rightarrow$  mixed high and normal AG acidosis

-6 to 6 only a high AG acidosis exists

over 6 mixed high AG acidosis and metabolic alkalosis