

Seven ABG steps, by Alex Flaxman, MD, MSE

1. Calc A-a gradient = $150 - [1.25 \times PaCO_2] - PaO_2$
2. Alk or acid
3. 1° disturb resp or metab
4. if resp, ac or ch

Expected pH chg

$$\begin{aligned} \text{ac resp acid pH decrease} &= .08(pCO_2 - 40) / 10 \\ \text{ch resp acid pH decrease} &= .03(pCO_2 - 40) / 10 \\ \text{ac resp alk pH increase} &= .08(40 - pCO_2) / 10 \\ \text{ch resp alk pH increase} &= .03(40 - pCO_2) / 10 \end{aligned}$$

Expected bicarb

$$\begin{aligned} \uparrow \text{ ac resp acid} & .1 (pCO_2 - \text{norm } pCO_2) \\ \uparrow \text{ ch resp acid} & .35 (pCO_2 - \text{norm } pCO_2) \quad \text{some say coefficient 0.4} \\ \downarrow \text{ ac resp alk} & .2 (\text{norm } pCO_2 - pCO_2) \\ \downarrow \text{ ch resp alk} & .5 (\text{norm } pCO_2 - pCO_2) \end{aligned}$$

Can also use

$$\begin{aligned} \text{Resp acidosis} & \text{ Ac for ea } 10 \uparrow \text{ of } pCO_2, HCO_3 \uparrow \text{ by } 1 \\ & \text{ Ch for ea } 10 \uparrow \text{ of } pCO_2, HCO_3 \uparrow \text{ by } 2 \\ \text{Resp alk} & \text{ Ac for ea } 10 \downarrow \text{ of } pCO_2, HCO_3 \downarrow \text{ by } 1 \\ & \text{ Ch for ea } 10 \downarrow \text{ of } pCO_2, HCO_3 \downarrow \text{ by } 5 \end{aligned}$$

$$\text{Metab acid } pCO_2 = (1.5 \times HCO_3) + 8 \pm 2$$

Another way: for ea 1 ↓ in bicarb from 24, pCO_2 ↑ by 1

$$\text{Metab alk} = \text{for ea } \uparrow \text{ in } HCO_3 \text{ by } 1, pCO_2 \uparrow \text{ by } \frac{1}{2}$$

5. If metab, is there an AG? $Na - (Cl + HCO_3)$
 - For every 1 g/dL decrease in albumin below 4, the AG should be raised by 2.5 mEq/L
 - (this is generally not on med school- or residency-level exams)
6. If AG metab acidosis exists, is there another metab proc going on? Use δ/δ or corrected HCO_3 .
7. $\delta/\delta = \text{chg in AG} / \text{chg in } HCO_3 = (AG - 12) / (24 - HCO_3)$

For med school- and residency-level exams:

 - $\delta/\delta > 1 \rightarrow$ concurrent metab alk
 - $\delta/\delta < 1 \rightarrow$ concurrent non-AG (hyperchloremic) metab acid

< 0.4 then there is a concomitant hyperchloremic normal anion gap acidosis
0.4-0.8 then consider combined high AG and normal AG acidosis but ratio is often < 1 in acidosis associated with renal failure
1-2 is usual for uncomplicated high AG acidosis
Lactic acidosis yields an average value of 1.6
DKA yields value closer to 1 b/c of urine ketone loss (espec if pt not dehydrated)
If > 1 or definitely 2 then concurrent metabolic alkalosis (maybe a pre-existing one?)

$$[\text{corrected } HCO_3 = \text{measured } HCO_3 + AG - 12]$$

- 6 \rightarrow mixed high and normal AG acidosis
- 6 to 6 only a high AG acidosis exists
- over 6 mixed high AG acidosis and metabolic alkalosis